

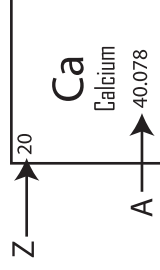
Name : \_\_\_\_\_

## Subatomic Particles

**Atomic Number:** An atomic number is the number of protons present in the atomic nucleus of an element. It is denoted by the letter "Z".

**Mass Number:** A mass number is the number of protons and neutrons present in the atomic nucleus of an element and is represented by the letter "A".

**Number of Protons**  
**Number of Neutrons**



Z → Atomic Number  
A → Mass Number

Complete the missing pieces of information in the table below.

Element	Atomic Number	No. of Neutrons	No. of Electrons
Be		5	4
	44	57	
B		6	
	8	8	
Cl		18	8
	10	10	
	78		

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